Appendix 3: A list of possible set-ups

Table S3. List of possible set-ups and working principles

Possible set-up	Working principles	Example(s)	Similar set-up
A	 The yeast-sugar extract is placed in a syringe The yeast carries out fermentation in the absence of oxygen Carbon dioxide is produced The pressure in the syringe increases/the volume of yeast-sugar extract in the syringe expands The increased pressure in the syringe pushes the plunger outwards The change in volume of the mixture in the syringe over time represents the rate of fermentation 	Knabb and Misquith (2006)	
B	 The yeast-sugar extract is placed in a burette The yeast carries out fermentation in the absence of oxygen Carbon dioxide is produced The carbon dioxide rises up and is trapped in the burette The change in volume of the mixture in the burette over time represents the rate of fermentation 	Collins and Bell (2004); Yurkiewicz, Ostrovsky, and Knickerbocker (1989)	20%

C	 The yeast-sugar extract is placed in a bottle The yeast carries out fermentation in the absence of oxygen Carbon dioxide is produced and is collected over water/forms bubbles under water The number of bubbles evolved/volume of carbon dioxide displaced over time represents the rate of fermentation 	Chan (2016); Weinberg (2018)	
D	 0.1 M sodium hydrogencarbonate is added to the yeast-sugar extract such that the resultant mixture is just above the end point of phenolphthalein The yeast carries out fermentation in the absence of oxygen Carbon dioxide is produced and dissolves in the yeast-sugar solution, forming carbonic acid This lowers the pH of the mixture The pink colour of the mixture fades when the end point of the indicator is reached The reciprocal of the time required to reach the end point represents the rate of fermentation 	Hong Kong Examinations and Assessment Authority (2009)	

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	Note the later	

- The yeast-sugar extract is placed in capped micro-centrifuge tubes with holes
- The yeast carries out fermentation in the absence of oxygen
- Carbon dioxide is lost
- This decreases the mass of the microcentrifuge tubes
- The change in mass over time represents the rate of fermentation

Grammer (2012)

