### e-Learning in Science classroom

12/10/2018

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Learning Management System

OOLOGY OF OUT DE	COURSES GROUPS RESOURCES
	2018-19 S1 Integrated Science: Whole Year
	😭 Add Materials 🔻 Options 🕶
Course Options	Science etv.xls 374 KB
Materials	Ø Unpublished
Updates	Science(S1-3)_supp_e_2017.pdf 3MB
Gradebook	<ul> <li>Unpublished</li> </ul>
Grade Setup	S1 IS TP 2018-2019.doc 165 KB
Radges	© Unpublished
Attendance	S1 IS TP 2018-2019 (19Sept2018).doc 234 k8
Access Code × 9NS2T-55429 Reset	> Dictation & Check Understanding
Information	> Chapter 1 Introducing Science
Grading periods 18-19 Term 1, 18-19 Term2	> 🚞 Chapter 2 Water
	> Chapter 3 Looking at Living Things

S schoology	PGRADE COURSES GROUPS RESOURCES	🔍 🎛 🗮 🖂 🖓 Eva Lam
Scence	2018-19 52 Integrated Science: Whole Year 7 Living things and air	▲ Prev Next ▶
<ul> <li>Course Options</li> </ul>	Add Materials   Options	
Materials •	> 7.1 Gases in the air	\$~
Gradebook	> 7.2 Air and burning	\$~
🙀 Grade Setup	> 7.3 How human obtain energy	\$~
Attendance	> 7.4 How green plants obtain energy	\$×
Access Code × MGD8X-D2M9G Reset	<ul> <li>7.5 Gaseous exchange in animals and plants</li> <li>Ø Unpublished</li> </ul>	\$~
Information Grading periods	16-17 S2 IS 1st UT Question Paper.pdf 409 кв Ø Unpublished	\$~
18-19 Term1, 18-19 Term2	16-17 S2 IS 1st UT Marking Scheme.pdf 152 KB Ø Unpublished	¢~
	<ul> <li>17-18 S2 IS 1st UT Question.pdf 879 K8</li> <li>Ø Unpublished</li> </ul>	\$~
	<ul> <li>17-18 S2 IS 1st UT Marking Scheme.pdf 16 KB</li> <li>Ø Unpublished</li> </ul>	\$-~
	> 🛅 Study Journal	\$-

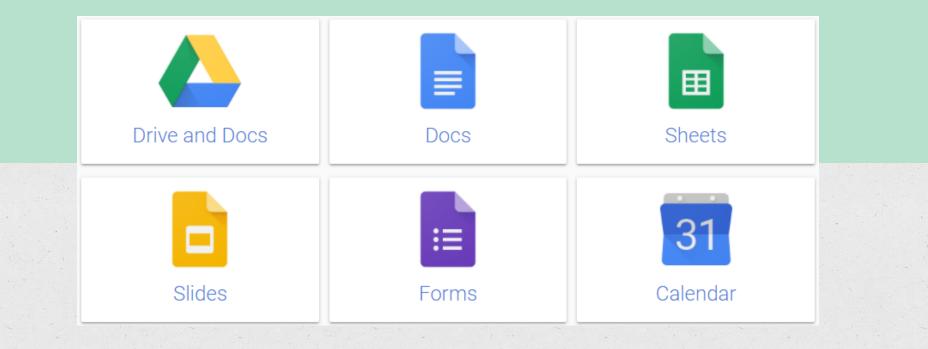
<b>c</b> ence	2018-19 S2 Integrated Science: Whole Year • 7 Living things and air 7.1 Gases in the air	Prev Next
Course Options	Add Materials • Options •	
Updates	<ul> <li>7.1 Gases in the air.ppt SMB</li> <li>Ø Unpublished</li> </ul>	\$~
Gradebook Grade Setup Badges	ETV - Air and Burning	\$~
Attendance Members	The Periodic Table of Videos - University of	\$ <b>~</b>
ccess Code X	Laboratory 7.1	\$~
IGD8X-D2M9G Reset	Laboratory 7.2	\$~
ading periods -19 Term1, 18-19 Term2	Laboratory 7.3	\$ <b>~</b>
	Laboratory 7.4	\$×
	Laboratory 7.5	\$~
	Carbon dioxide test with lime water	\$~
	7.1 Gases in the air.pdf 4MB	÷~



#### What Is Nearpod?

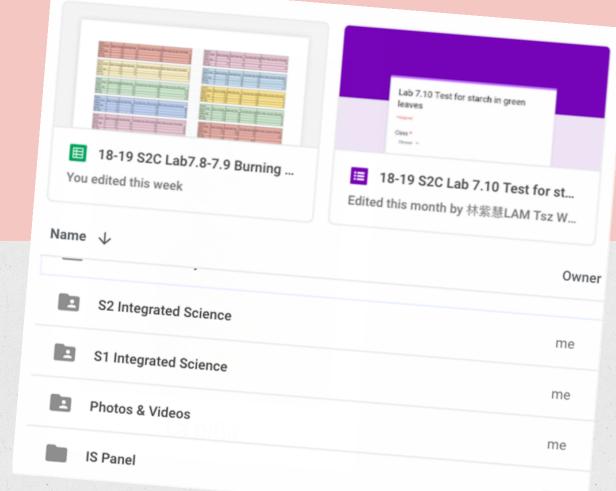
https://youtu.be/g3GVkM\_GVa4

An interactive classroom tool





Google drive



← 18-19 S2C Lab 7.3-7.5 Diff	erences between breathed air and unbr 🛕 ☆ 🗭	
	QUESTIONS RESPONSES 10	
	Lab 7.3 6a. In which jar does the burning splint burn for a longer time? * <ul> <li>Breathed air</li> <li>Unbreathed air</li> </ul>	Google Forms
	Lab 7.3 6b. What does this result show? * Short-answer text	
	Lab 7.3 7. Put some hydrogencarbonate indicator into a jar of breathed air. * What happens to the colour of the indicator?	
	Lab 7.3 8a. Put some hydrogencarbonate indicator into a jar of unbreathed * air. What happens to the colour of the indicator?	

	18-19 S2C Lab 7.3-7.5 Differences between breathed air and unbreathed air (Responses)       Image: Share in the second seco						
5	• → ➡ ➡ 100% - £ % .0 .0 123 - Arial - 10	- B I S A À 🖽 🗄	E → <u></u> ± → I÷ → Þ	- σ - μ γ - Σ - Λ			
fx							
	A B C D E F	G	н	1			
1	Times Class Class N Lab Lab 7.3 6b. What does this result show?	Lab 7.3 7. Put some hydrogencarbonate	Lab 7.3 8a. Put some hydro	Lab 7.3 8b. What does the above result show?			
2	24/09/; S2C 9 19 CI Unbi Breathed air contains less oxygen	It changes from red to yellow	There is no colour change	Breathed air contains more carbon dioxide than unbreathed air			
3	24/09/22C 5, 15 Gi Unbi Oxygen in unbreathed air is more than that in breathed	a It turns to yellow	It turns to purple	Carbon dioxide in breathed air is more than that in unbreathed a			
4	24/09/2 2c 3,13, Ja Unbi Unbreathed air have more oxygen	Turned into yellow colour	Turn to a little bit yellow	Breathed air have more carbon dioxide			
5	24/09/2 S2C 8, 18 CI Unbi Breathed air contains less oxygen than unbreathed air	Yellow	Red	Breathed air contains more Carbon dioxide than unbreathed air			
6	24/09/22C 10,2C R Unbi Breathed air contains less oxygen than unbreathed air	Turns yellow	Remains red	Breathed air contains more carbon dioxide than unbreathed air			
7	24/09/2 S2C 6, 16 Al Unbi There are more oxygen in unbreathed air	Yellow	Red	There are more carbonate dioxide in breathed air			
8	24/09/2 S2C 1, 11, C( Unbi Unbreathed air contains more oxygen than breathed air	The indicator turned from red to yellow.	The indicator stayed in red.	Breathed air comtains more carbon dioxide than unbreathed air.			
9	24/09/2 2c 4 14 Ar Unbi Breathed air contains less oxygen than breathed air	Yellow	Red	Breathed air contains more carbon dioxide than unbreathed air			
10	24/09/; 2C 2 12 C Unb There are more oxygen in the unbreathed air	It turns yellow from red	Remains red	There are more carbon dioxide in breathed air			
11	24/09/; 2C 7 17 C Unbi Unbreathed air have more oxygen	It turns to yellow	Remain unchange	Breathed air have more carbon dioxide			
12							
13							



■ 18-19 S2C Lab7.8-7.9 Burning of food ☆ ■ File Edit Was 2 days ago					
5		100% - £	% .0 <sub>_</sub> .0 <u>_</u> 123	- Arial - 10 -	B I S A 🍝 🖽 EE -
fx	Group 1				
	А	В	С	D	E
2	Type of snack	Mass of snack(g)	Initial water temp.	Final water temp. after the snack	Rise in temp. per gram of the snack
3	Peanut	0.4	25	37	30
4	Potato chips	0.5	25	42	34
5	Biscuits	0.4	25	50	62
6	Dry meat	0.5	25	46	42
7					
8	Group 2				
9	Type of snack	Mass of snack(g)	Initial water temp.	Final water temp. after the snack	Rise in temp. per gram of the snack
10	Peanut	0.5	25	42	34
11	Potato chips	0.5	25	30	10
12	Biscuits	0.5	25	28	6
13	Dry meat				
14					
15	Group 3				
16	Type of snack	Mass of snack(g	Initial water temp.	Final water temp. after the snack	Rise in temp. per gram of the snack
17	Peanut	0.4	25	48	57.5
18	Potato chips	0.4	26	49	57.5
19	Biscuits	0.5	26	41	30
20	Dry meat	0.4	26	43	42.5
21					
22	Group 4				
23	Type of snack	Mass of snack(g	Initial water temp.	Final water temp. after the snack	Rise in temp. per gram of the snack
24	Peanut	0.4	25	54	29



Google Sheets

# I. IODINE TEST

TRANSFER A FEW DROPS OF DISTILLED WATER AND STARCH WATER TO A WELL ON A SPOT PLATE RESPECTIVELY.

LABEL THE LIQUID ON THE SPOT PLATE AND TAKE A PHOTO.



# I. IODINE TEST

ADD A DROP OF IODINE SOLUTION TO THE WELL OF DISTILLED WATER AND STARCH WATER RESPECTIVELY. OBSERVE THE COLOUR CHANGE.

-THE IODINE SOLUTION TURNS BROWN WHEN WE ADDED IT INTO DISTILLED WATER.

THE IDDINE SOLUTION TURNS BLUE-BLACK WHEN WE ADDED IT INTO STARCH WATER.





. IODINE SOLUTION CHANGES FROM BROWN TO BLUE-BLACK IN COLOUR WHEN STARCH IS PRESENT.

## II. TEST FOR STARCH IN

GREEN LEAVES

REMOVE A GREEN LEAF FROM A POTTED PLANT. TAKE A PHOTO OF THE LEAF BEFORE IODINE TEST.

#### 1. BOIL THE LEAF IN WATER

BOIL ISO ML WATER. PUT THE LEAF INTO THE BEAKER OF BOILING WATER FOR 4 MINUTES.

WHAT IS THE PURPOSE OF PUTTING THE LEAF INTO THE BOILING WATER?

. TO DESTROY THE CELL MEMBRANES OF THE LEAF CELLS.



### 2. SOAK THE LEAF IN HOT ALCOHOL

- . WHAT HAPPEN TO THE COLOURS OF THE LEAF AND THE ALCOHOL?
- . IT TURNS TO PALE YELLOW .
- . WHAT IS THE PURPOSE OF PUTTING THE LEAF INTO ALCOHOL?
- TO REMOVE THE CHLOROPHYLL FROM THE LEAF.



## 3. WASH THE LEAF IN HOT WATER

· Observe the colour of the leaf. Take a photo of it.

. WHAT IS THE PURPOSE OF PUTTING THE LEAF IN HOT WATER AGAIN?

. To wash away the Alcohol and soften the Leaf.



### 4. ADD IODINE SOLUTION TO THE LEAF

· SPREAD THE LEAF ON A WHITE TILE. ADD A FEW DROPS OF IODINE SOLUTION TO THE LEAF.

. WHAT HAPPENS TO THE COLOUR OF THE LEAF?

. IT TURNS TO BLUE-BLACK.

. WHAT DOES THIS RESULT SHOW?

. THERE IS STARCH IN THE LEAF.



Edmodo	Nearpod		Scratch	
Schoology	<u>Quizizz</u>		Micro:bit	Mainly
Google Docs	Kahoot!	eAssessment	littleBits	for STEM
Google Drive	<u>Socrative</u>		mBlocky	STEM
Google Sheets	Notability		Tinkercad	
Google form Record	iMovie		Desmos	
Keynote learning progress	Explain Every	<u>/thing</u>	Geogebra	Mainly
Numbers	<u>Skitch</u>	Make/ Edit photos/	Slice it	for Maths
Pages	Edpuzzle	video	<u>SymShuffle</u>	